

The Student Volume 5



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Reviews

I actually began reading this article pdf. It really is filled with wisdom and knowledge You wont sense monotony at at any time of the time (that's what catalogues are for concerning should you request me).

(Ena Klein MD)

THE STUDENT VOLUME 5



RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 24 pages. Original publisher: Greenbelt, Md. : NASA Goddard Space Flight Center, 2000 OCLC Number: (OCoLC)83756137 Excerpt: . . . 9. 3. 2 Calculated Variables Porewater CH₄ concentration (PWCH₄) was calculated as: $CH_4hs \cdot P \cdot Po \cdot Kh \cdot Vhs \cdot Vw \cdot (CH_4hs - CH_4amb) \cdot P \cdot RT$ where: CH₄hs is the equilibrated headspace concentration of CH₄ (ppm) P and Po are ambient and sea-level atmospheric pressures (101325 Pa) CH₄ Kh is the Henrys law constant for at ambient temperature (298 K) Vhs and Vw are the volumes of headspace and water in the syringe CH₄amb is the ambient CH₄ concentration (1.8 ppmv) R is the universal gas constant (8314.4 L Pa K mol T is the ambient temperature (298 K) Porewater concentration calculated as: (PWCO₂) was CO₂ CO₂hs P Po Kh (I Ka1 10pH (I Ka2 10pH)) Vhs Vw (CO₂hs-CO₂amb) P RT where: is the equilibrated headspace concentration of (ppm) CO₂hs CO₂ P and Po are ambient and sea-level atmospheric pressures (101325 Pa) Kh is the Henrys law constant for CO₂ at ambient temperature (298 K) Ka1 and Ka2 are the 1st and 2rid dissociation constants for aqueous CO₂ pH is the pH of the porewater after stripping Vhs and Vw are the volumes of headspace and water in the syringe CO₂amb is the ambient CO₂ concentration (350 ppmv) R is the universal gas constant (8314.4 L Pa K tool) T is the ambient temperature (298 K) 9. 4 Graphs and Plots None given. 10. Errors 10. 1 Sources of Error During sampling, a leak in...



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